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Code: 021307

B.Tech. 3rd Semester Exam., 2013

THERMODYNAMICS

Time: 3 hours Full Marks: 70

Instructions:

- (i) All questions carry equal marks.
- (ii) There are NINE questions in this paper.
- (iii) Attempt FIVE questions in all.
- (iv) Question No. 1 is compulsory.
- (v) Use of Tables and Charts permitted.
- 1. State whether the following statements are True or False (any seven):
 - (a) The sum of internal energy and pressure volume product is called enthalpy.
 - (b) Diesel cycle consists of two constant volumes and two adiabatic processes.
 - (c) For the same compression ratio and same heat input the thermal efficiency of Otto cycle is less than that of Diesel cycle.
 - (d) Intensive properties are independent of the mass of the system.

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(2)

- (e) An isolated system is one which permits the passage of energy only.
- (f) Liquids have two specific heats.
- (g) Thermal power plant works on Rankine cycle.
- (h) Sublimation is the process of changing from solid state to direct gas state.
- (i) On psychrometric chart, DBT lines are horizontal.
- (j) During sensible cooling process, specific humidity decreases.
- (a) State and explain zeroth law of thermodynamics.
 - (b) Derive an expression of displacement work for an adiabatic process.
 - (c) What do you mean by flow work?
- (a) Show that the efficiency of all reversible heat engines operating between the same temperature levels is the same.
 - (b) What is a Carnot cycle? Derive its efficiency with the help of p-V diagram and block diagram.

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- 4. 2 kg of ice at -6 °C is exposed to the atmosphere which is at 25 °C. The ice melts and comes into thermal equilibrium with the atmosphere. Determine—
 - (a) the entropy increase of the universe;
 - (b) the minimum amount of work necessary to convert the water back into ice at -6 °C.
- 5. (a) What do you mean by triple point?
 - (b) A large insulated vessel is divided into two compartments, one containing 5 kg of dry saturated steam at 2 bar and the other 10 kg of steam, 0.8 dry at 5 bar. If the partition is removed and the steam is mixed thoroughly and allowed to settle, find the final pressure, steam quality and entropy change.
- 6. (a) What is the reversible cycle that represents the simple steam power plant? Draw the flow, p-V, T-S and h-S diagrams of this cycle.
 - (b) What do you understand by the mean temperature of heat addition?
 - (c) How are the maximum temperature and maximum pressure in the Rankine cycle fixed?

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- 7. (a) Develop an expression for the thermal efficiency of an air-standard Diesel cycle.
 - (b) Compare the efficiency of Otto, Diesel and dual cycles in the following cases with the help of p-V and T-S diagrams:
 - (i) For the same compression ratio
 - (ii) For the same maximum pressure and temperature
- 8. 3 kg of air at 40 °C DBT and 20 °C WBT is adiabatically mixed with 2 kg of air at 25 °C DBT and 10 °C DPT. Determine the final specific humidity and temperature of the mixture without using psychrometric chart.
- 9. (a) A mixture of ideal gases consists of 2 kg of O₂ and 4 kg of CO₂ at a pressure of 3 bar and temperature of 25 °C. If the mixture is heated at constant pressure to 50 °C, find the change in entropy of the mixture.
 - (b) Write the Maxwell's equations.

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